# **Engineering Mathematics**

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Mathematics 3321



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## Section 2.4.2 Exponential Growth and Decay

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The differential equation

$$A'(t) = rA(t)$$
 for all t in J (1)

indicates that the rate of change in A at time t is proportional to the amount A(t). The interval J is usually  $[0, \infty)$  and we will take it to be that. The equation can be solved by writing it in the equivalent form

$$A'(t) - rA(t) = 0$$

which is a first order linear differential equation. It can also be solved by writing it in the equivalent form (assuming that  $A(t) \neq 0$ )

$$\frac{A'(t)}{A(t)} = r$$

which is a separable differential equation.

Either way, we find that A is a solution to

$$A'(t) = rA(t)$$
 for all t in J (1)

if and only if

$$A(t) = ce^{rt}$$

for some number C and all t in J. Setting t = 0, it follows that

$$A(0) = C$$

so

$$A(t) = A(0)e^{rt} \text{ for all } t \text{ in } J.$$
(2)

Equation (2) gives the solution to (1) in terms of A(0) which is called the initial amount.

### GROWTH

When r > 0, A is said to exhibit exponential growth and r is called the growth constant. In this case, the doubling time is the number D such that

A(D)=2A(0).

From (2), letting t = D, we see that

$$2A(0) = A(0)e^{rD}$$

Thus (assuming that  $A(0) \neq 0$ )

$$2 = e^{rD}$$

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$$\ln 2 = rD$$

$$D = \frac{\ln 2}{r}$$

and

The doubling time also has the following property. When  $t_0$  is any number in J,

$$A(t_0+D)=2A(t_0).$$

To see this note that

$$\begin{aligned} A(t_0+D) &= A(0)e^{r(t_0+D)} = A(0)e^{rt_0}e^{rD} \\ &= A(0)e^{rD}e^{rt_0} = 2A(0)e^{rt_0} = 2A(t_0) \end{aligned}$$

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#### DECAY

When r < 0, A is said to exhibit exponential decay and -r is called the decay constant. Let

$$k = -r$$
 so that  $r = -k$ .

Equation (2) becomes

$$A(t) = A(0)e^{-kt} \text{ for all } t \text{ in } J.$$
(3)

(B)

In this case, the half-life is the number H

$$A(H)=\frac{1}{2}A(0).$$

From (3), letting t = H, we see that

$$\frac{1}{2}A(0) = A(0)e^{-kH}$$

Thus (assuming that  $A(0) \neq 0$ )

$$\frac{1}{2} = e^{-kH}$$

so

$$\ln \frac{1}{2} = -kH$$

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so 
$$H = \frac{\ln 2}{k}$$
and 
$$k = \frac{\ln 2}{H}.$$

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The half-life also has the following property. When  $t_0$  is any number in J,

$$A(t_0+H)=\frac{1}{2}A(t_0).$$

To see this note that

$$\begin{aligned} A(t_0 + H) &= A(0)e^{-k(t_0 + H)} = A(0)e^{-kt_0}e^{-kH} = A(0)e^{-kH}e^{-kt_0} \\ &= \frac{1}{2}A(0)e^{-kt_0} = \frac{1}{2}A(t_0) \end{aligned}$$

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### EXAMPLES

What is the half-life of a radioactive substance if it takes 5 years for one-third of the material to decay?

Solution:

$$A(t) = A(0)e^{-kt}$$

SO

$$A(5) = A(0)e^{-5k}$$

After 5 years two-thirds of the material remains so

$$A(5)=\frac{2}{3}A(0).$$

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Thus

$$\frac{2}{3}A(0) = A(0)e^{-5k}.$$

Dividing each side by A(0) then taking In of each side we have

$$\ln\frac{2}{3} = -5k.$$

Thus

$$k = -\frac{\ln(2/3)}{5} = \frac{\ln 3 - \ln 2}{5}.$$
  
Since the half-life is  $\frac{\ln 2}{k}$ , it follows that
$$H = \frac{5\ln 2}{\ln 3 - \ln 2} \approx 8.548 \text{ years}$$

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**Additional Examples:** See Section 2.4.2 of the text and the notes presented on the board in class.

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**Suggested Problems**. Do the odd numbered problems for section 2.4.2. The answers are posted on Dr. Walker's web site.

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