## Math 3331 Differential Equations 2.2 Solutions to Separable Equations

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### 2.2 Solutions to Separable Equations

#### • General Method: Separation of Variables

- Separable Equation
- Exponential Equation
- General Method
- Explicit Solution
- Implicitly Defined Solutions
- Applications
  - Radioactive Decay and Half-life
  - Newton's Law of Cooling
- Worked out Examples from Exercises:
  - Find General Solutions: 1, 3, 5, 9, 11
  - Find Solutions to IVPs and IoEs: 13, 15, 17, 19
  - Application: 26, 33



# The General Method: Separation of Variables

Form: 
$$\frac{dy}{dt} = g(t)f(y)$$

### Implicit Solution:

$$[1/f(y)]dy = g(t)dt$$
$$\int [1/f(y)]dy = \int g(t)dt \quad (*)$$
or  $H(y) = G(t) + C$  where
$$H(y) = \int [1/f(y)]dy$$
$$G(t) = \int g(t)dt$$

Solve (\*) for  $y \rightarrow$  explicit solution **Note:** (\*) may have several solutions. Use IC to choose the right one.

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# Example

Ex.: 
$$\frac{dy}{dt} = ty^2$$
  
 $(1/y^2)dy = t dt$   
 $\Rightarrow \int (1/y^2)dy = \int t dt$   
 $\Rightarrow -1/y = t^2/2 + C$   
 $\Rightarrow y(t) = -1/(t^2/2 + C)$   
 $= -2/(t^2 + 2C)$ 



### Example: Exponential Equation

Ex.: Find gen. sol. to 
$$dx/dt = rx$$
  
 $\frac{dx}{x} = r dt \Rightarrow \ln |x| = rt + C$   
 $\Rightarrow |x(t)| = e^{rt+C} = e^C e^{rt}$   
 $x(t) > 0 \Rightarrow x(t) = e^C e^{rt}$   
 $x(t) < 0 \Rightarrow x(t) = -e^C e^{rt}$   
Set  $A = e^C$  if  $x > 0$ ,  $A = -e^C$  if  $x < 0$   
 $\Rightarrow x(t) = Ae^{rt}$   
with arbitrary constant  $A$  (can be 0)

Initial value: x(0) = A



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#### Example: IVP

**Example:** general linear equation with constant coefficients y' = ry + a, IC:  $y(0) = y_0$  ( $r, a, y_0$ : arbitrary parameters)  $[1/(ry + a)]dy = dt \Rightarrow (\ln |ry + a|)/r = t + C \Rightarrow |ry + a| = e^{rt+rC} = e^{rC}e^{rt}$   $\Rightarrow ry + a = Ae^{rt}$  ( $A = \pm e^{rC}$ )  $\Rightarrow y(t) = (Ae^{rt} - a)/r$ Invoke IC:  $y(0) = (A - a)/r = y_0 \Rightarrow A = ry_0 + a \Rightarrow y(t) = (y_0 + a/r)e^{rt} - a/r$ 



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### Implicitly Defined Solutions

Find sols. of 
$$x' = 2tx/(1+x)$$
  
s.t.  $x(0) = 1$ ,  $x(0) = -2$ , and  $x(0) = 0$ .

Answer:

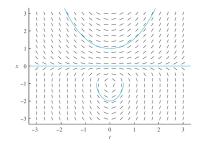
$$(1+1/x)dx = 2tdt, \quad x \neq 0$$
  

$$\Rightarrow x + \ln(|x|) = t^{2} + C$$
  
(i) For  $x(0) = 1$   

$$\Rightarrow C = 1$$
  

$$x + \ln x - 1 = t^{2}$$
  

$$\Rightarrow x(t) \text{ implicitly defined}.$$





Ex. 1: 
$$y' = xy$$
  
 $(1/y)dy = xdx \Rightarrow \ln |y| = x^2/2 + C \Rightarrow |y| = \exp(x^2/2 + C) = e^C e^{x^2/2}$   
 $\Rightarrow y(x) = Ae^{x^2/2}, A = e^C \text{ or } A = -e^C$ 



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# **Ex. 3:** $y' = e^{x-y}$ $e^y dy = e^x dx \Rightarrow e^y = e^x + C \Rightarrow y(x) = \ln(e^x + C)$



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**Ex. 5:** 
$$y' = y(x+1)$$
  
 $(1/y)dy = (x+1)dx \Rightarrow \ln |y| = x^2/2 + x + C \Rightarrow |y| = e^C e^{x+x^2/2} \Rightarrow y(x) = Ae^{x+x^2/2}$ 



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Math 3331 Differential Equation

**Ex.** 9: 
$$x^2y' = y \ln y - y' \Rightarrow y' = (y \ln y)/(1 + x^2)$$
  
 $[1/(y \ln y)]dy = [1/(1 + x^2)]dx \Rightarrow \ln(\ln y) = \arctan x + C$   
 $\Rightarrow y(x) = \exp(e^C e^{\arctan x}) = \exp(De^{\arctan x}) \quad (D = e^C)$ 



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**Ex. 11:** 
$$y^3y' = x + 2y' \Rightarrow y' = x/(y^3 - 2)$$
  
 $(y^3 - 2)dy = x dx \Rightarrow y^4/4 - 2y = x^2/2 + C \Rightarrow \text{ implicit sol.: } y^4 - 8y - 2x^2 = D \ (D = 4C)$ 



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$$\begin{aligned} & \mathsf{Ex. 13:} \ y' = y/x, \ \mathsf{IC:} \ y(1) = -2 \\ & \text{General sol.:} \qquad (1/y)dy = (1/x)dx \Rightarrow \ln|y| = \ln|x| + C \\ & \Rightarrow \ |y| = \exp(C + \ln|x|) = e^C e^{\ln|x|} = e^C |x| \Rightarrow y(x) = Ax \ (A = \pm e^C) \\ & \text{Match } C \ \text{to IC:} \qquad y(1) = A = -2 \Rightarrow y(x) = -2x; \ \mathsf{IoE:} \ (0,\infty) \end{aligned}$$



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Ex. 15: 
$$y' = (\sin x)/y$$
, IC:  $y(\pi/2) = 1$   
 $y \, dy = \sin x \, dx \Rightarrow y^2/2 = -\cos x + C \Rightarrow y = \pm \sqrt{D - 2\cos x} \quad (D = 2C)$   
 $y(\pi/2) = 1 > 0 \Rightarrow \text{need}'+'-\text{sign} \Rightarrow y(\pi/2) = \sqrt{D} = 1 \Rightarrow y(x) = \sqrt{1 - 2\cos x}$   
Find IoE: need  $\cos x < 1/2 \Rightarrow \text{IoE:} \quad (\pi/3, 5\pi/3)$ 



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**Ex. 17:** 
$$y' = 1 + y^2$$
, IC:  $y(0) = 1$   
 $[1/(1+y^2)]dy = dt \Rightarrow \arctan y = t + C \Rightarrow y = \tan(t+C) + k\pi \ (k: \text{ integer})$   
Since  $y(0) = 1 \Rightarrow k = 0 \Rightarrow y(t) = \tan(t+C)$   
Invoke IC:  $y(0) = \tan C = 1 \Rightarrow C = \pi/4 \Rightarrow y(t) = \tan(t + \pi/4)$   
For IoE: need  $t + \pi/4 > -\pi/2$  and  $t + \pi/4 < \pi/2 \Rightarrow \text{IoE:} (-3\pi/4, \pi/4)$ 



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**Ex. 19:** 
$$y' = x/y$$
, IC<sub>1</sub>:  $y(0) = 1$  and IC<sub>2</sub>:  $y(0) = -1$   
 $y \, dy = x \, dx \Rightarrow y^2/2 = x^2/2 + C \Rightarrow y = \pm \sqrt{x^2 + D} \quad (D = 2C)$   
IC<sub>1</sub>:  $y(0) = 1 \Rightarrow y(0) = +\sqrt{D} = 1 \Rightarrow y(x) = \sqrt{1 + x^2}$   
IC<sub>2</sub>:  $y(0) = -1 \Rightarrow y(0) = -\sqrt{D} = -1 \Rightarrow y(x) = -\sqrt{1 + x^2}$ 



### Radioactive Decay

#### N(t): $\sharp$ of radioactive atoms

• Model:  $dN/dt \sim -N$ 

 $\Rightarrow dN/dt = -\lambda N$ 

• Solution: 
$$N(t) = N_0 e^{-\lambda t}$$

• Half-life:

$$N(t)/N(0) = e^{-\lambda t} = 1/2$$

 $\Rightarrow t = (\ln 2)/\lambda \equiv T_{1/2}$ 

• Natural log of ratios:

 $\ln[N_0/N(t)] = \lambda t$ 

- Use  $\lambda = (1/t) \ln[N_0/N(t)]$  to determine  $\lambda$  from measurement
- Use  $t = (1/\lambda) \ln[N_0/N(t)]$ to determine time  $t^*$  s.t.  $N(t^*) = N^*$  for given  $N^*$

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**Ex. 25:** After t = 4 hrs, 80 mg of a 100 mg sample of Tritium remain. Determine  $\lambda$  and  $T_{1/2}$ .

Answer:  $\lambda = (1/4) \ln[100/80] = 0.056/hrs$  $T_{1/2} = (\ln 2)/0.056 = 12.43 hrs$ 



**Ex. 26:**  $T_{1/2} = 6 hrs$  for Technitium 99*m*. What remains after 9 hrs if  $N_0 = 10 g$ ?

Answer:  $\lambda = (\ln 2)/6 = 0.116/hr$  $\Rightarrow N(9) = 10e^{-0.116 \times 9} = 3.54 g$ 



### Newton's Law of Cooling

- T(t): temperature of objet
- A: surrounding temperature
  - Model:  $dT/dt \sim A T$

$$\Rightarrow dT/dt = k(A - T)$$

• Solution (see Example p.5):

$$T(t) = A + e^{-kt}(T_0 - A)$$

• 
$$(T-A)/(T_0-A) = e^{-kt} \Rightarrow$$

$$kt = \ln[(T_0 - A)/(T(t) - A)]$$

$$k = (1/t) \ln[(T_0 - A)/(T(t) - A)]$$
  

$$\rightarrow \text{determine } k$$

$$t = (1/k) \ln[(T_0 - A)/(T(t) - A)]$$
  

$$\rightarrow \text{ determine } t$$



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**Ex. 33:** Dead body found at t = 0 (midnight). Temperature  $31^{\circ}C$ .

1 hr later (1 am): Temperature 29<sup>o</sup>C Surrounding temperature:  $A = 21^{\circ}C$ *Question:* When did death (murder) occur?

Answer: t = 1 hr,  $T_0 = 31$ , T(1) = 29  $\Rightarrow k = (1/1) \ln[(31 - 21)/(29 - 21)]$ = 0.223/hr

Determine time at which T = 37:  $t = (1/k) \ln[(31 - 21)/(37 - 21)]$ = -2.11 hrs = -2 hrs 7 min

 $\Rightarrow$  Death occured at 9 : 53 pm

